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Test: Chemistry	**Class: 9th**
Time: 40 Min.	**Max. Marks: 20**
Name: _____ Date:	**Section:**

Q. No.1. Choose the Correct Answer. **1 x 5 = 5**

1. Chemical formula of rust is.

a) Fe_2O_3 b) $Fe_3O_4.\ 2H_2O$ c) $Fe(OH)_3$ d) $Fe_2O_3.\ nH_2O$

2. Stainless steel contains.

a) Nickle b) Iron c) Chromium d) All

3. The percentage of sliver in sterling silver is.

a) 92.5 b) 44.5 c) 22.5 d) 99.5

4. In Nelson's cell, cathode is made up of.

a) Iron b) Zinc c) Grphite d) Steel

5. Which of following is not an electrolytic cell?

a) Down's cell b) Galvanic cell c) Nelson d) None

(SUBJECTIVE TYPE)

Q. No.2. Give Short Answers of the following questions. **2 x 5 = 10**

1. What is the difference b/w rusting and corrosion?

2. What is meant by galvanizing?

3. What is the principle of electroplating?

4. Explain the role of O2 in rusting.

5. What is salt bridge? Give its function.

Q. No.3. Write note on electroplating of silver. **5**

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